Cambridge Checkpoint	Cambr Cambrid	idge In dge Che	t ernatio eckpoint	nal Ex	aminations				
CANDIDATE NAME									
CENTRE NUMBER						CANDIDATE NUMBER			
SCIENCE								11	13/02
Paper 2						For	Examinat	ion fron	n 2014
SPECIMEN PA	PER							45 m	inutes
Candidates ans	swer on th	e Questi	on Paper.						
Additional Mate	erials:	Pen Pencil Ruler			Calculator				

READ THESE INSTRUCTIONS FIRST

Write your Centre number, candidate number and name on all the work you hand in.Write in dark blue or black pen.You may use a soft pencil for any diagrams, graphs or rough working.Do not use staples, paper clips, highlighters, glue or correction fluid.

Answer **all** questions. You should show all your working in the booklet.

At the end of the examination, fasten all your work securely together. The number of marks is given in brackets [] at the end of each question or part question. The total number of marks for this paper is 50.

This document consists of 16 printed pages.

1 (a) The boxes show some parts of plants and the function of the parts.

Draw a line from each **part** to match its **function**.



(b) Complete the word equation for photosynthesis.



2 Complete the table by deciding if the information indicates that the element is a metal or a non-metal.

element	magnetic	state at room temperature	Does it conduct electricity?	metal or non-metal
А	no	liquid	yes	
В	yes	solid	yes	
С	no	liquid	no	
D	no	gas	no	
E	no	solid	yes	

3 Complete each explanation using these words.

area	force	pressure

(a) People who walk across snow may wear snowshoes.



The person does not sink	into the snow because the		acts on
a larger	so that the	is less.	[1]

(b) A sharp knife cuts through cheese more easily than a blunt knife.



The edge of the sharp knife has a smaller so the	
acting on the knife produces a larger	[1]

4 Plants and green algae need mineral salts to grow.

One mineral salt is magnesium sulfate.



Ahmed and Safia investigate the growth of algae.

They put different concentrations of magnesium sulfate solution into five flasks, A, B, C, D and E.

They then add the algae.



(c) Here are their results.

flask	concentration of magnesium sulfate (1 = dilute, 5 = most concentrated)	colour of algae (1 = light green, 10 = dark green)
Α	1	4
В	2	5
С	3	8
D	4	10
E	5	1

Complete the sentence to describe the pattern of their results.

As the concentration of magnesium sulfate increases from 1 to 5, the colour of the algae

[2]

(d) When the colour of the algae is dark green it has grown the most.

Complete the sentence.

Choose a word from the list.

carbon dioxide	fastest	the same	slowest	sugar	water

When the concentration of magnesium sulfate is 4 the algae grow _____. [1]

5 Look at the table.

It shows the melting points and boiling points of some elements.

element	melting point in °C	boiling point in °C
gold	1064	2856
iron	1538	2861
mercury	-39	357
oxygen	-219	-183
sodium	98	883

Use the table to answer these questions.

(a) Write down the name of the element that melts at 1538 °C.

[1]

(b) Which element is a liquid at room temperature (20 °C)?

[1]

(c) Which element is a non-metal?

[1]

- 6 Here are some statements about the solar system.
 - **A** The Earth orbits the Sun.
 - **B** The Earth spins on its axis.
 - **C** The Moon orbits the Earth.
 - **D** The stars orbit the Sun.
 - **E** The Sun orbits the Earth.
 - (a) Write down the letter of the statement that answers each of these questions.
 - (i) Why does the Sun appear to move across the sky each day?

[1]

- (ii) Why do some stars appear in the summer night sky but not in the winter night sky?

(iii) What did Copernicus and Galileo think was wrong?

[1]

(b) Venus is **not** a source of light.

Explain why it is possible to see Venus in the night sky.

.....[1]

7 The diagram shows a palisade cell.



(a) Which three structures, labelled in the diagram, are **not** found in animal cells?

	1	
	2.	
	3.	[3]
(b)	Name the part of the cell in which photosynthesis takes place.	
		[1]
The	e Earth is made up of three layers including the core and the crust.	
(a)	What is the name of the other layer?	
		[1]
(b)	The core is made up of mainly two metals.	
	One of these metals is nickel.	
	What is the name of the other metal?	
		[1]

8

(c) What is the approximate age of the Earth?

Tick (\checkmark) the correct box.



9 The diagram shows the energy flow into and out of a coal-fired power station.



(a) How much useful electrical energy is provided by the power station?

- MJ [1]
- (b) Calculate how much energy is wasted from the cooling tower?
 - _____ MJ [1]

[1]

10 Manjit puts a metal saucepan of milk on a cooker.



Thermal (heat) energy can be transferred by conduction, convection and radiation.

(a)	Which is the main process that transfers thermal energy through the milk?	
		[1]
(b)	Which is the main process that transfers thermal energy through the metal saucepan ?	[1]
		r.1

11 The alimentary canal consists of many different organs.

Look at the diagram of the alimentary canal.



(a)	What is the name of part A?	
		[1]
(b)	The stomach is not labelled on the diagram.	
	Draw a label line on the diagram to show the stomach.	[1]

(c) Draw lines to match the organ with its function.



12 Yuri does an experiment on sound.



Yuri connects a microphone to an oscilloscope.

He places the microphone at different distances from the loudspeaker.

He records the amplitude of the wave on the oscilloscope.

Here are some of the results on the oscilloscope.



(a) Use these pictures to complete the results table.

distance in cm	amplitude of wave in number of squares
4	2.5
6	
11	
25	1

Tick (\checkmark) the **correct** answer.

the amplitude does not change with distance	
the amplitude decreases with distance	
the amplitude increases with distance	
there is no pattern in the results	[1]

13 Hydrogen peroxide is used to make oxygen in the laboratory.

Hydrogen peroxide breaks down to form water and oxygen.

(a) This reaction is much faster when a chemical called manganese(IV) oxide is added to the hydrogen peroxide.

The manganese(IV) oxide is unchanged at the end of the reaction.

What type of chemical is manganese(IV) oxide?

Tick (\checkmark) the **correct** answer.



[1]

- (b) Blessy investigates the effect of temperature on the breakdown of hydrogen peroxide.
 - (i) Write down the variable she should **change**.

.....[1]

(ii) Write down one variable she should control (keep the same).

[1]

14 Some metal railings have started to rust.



(a) Which metal were the railings made of?

Circle the correct answer.

	aluminium	copper	iron	lea	d	[1]		
(b)) Which two substances must be present for the railings to rust?							
	Circle the two correct answers.							
	carbon dioxide	hydrogen	nitrogen	oxygen	water	[2]		
(c)) How can you prevent railings from rusting?							
						[1]		

15 This is a question about forces.

Look at the diagrams.



and

.....

[2]

16 A car is driving along a road.



Complete the sentences about the car.

Use words from the list

air resistance	chemical	elastic
friction	gravity	kinetic

The car is slowing down. The forces that slow the car down are

and _____.

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